

Evidence for a Proton–Protein Symport Mechanism in the Anthrax Toxin Channel

Daniel Basilio,¹ Stephen J. Juris,² R. John Collier,² and Alan Finkelstein¹

¹Department of Physiology and Biophysics, and Department of Neuroscience, Albert Einstein College of Medicine, Bronx, NY 10461

²Department of Microbiology and Molecular Genetics, Harvard Medical School, Boston, MA 02115

The toxin produced by *Bacillus anthracis*, the causative agent of anthrax, is composed of three proteins: a translocase heptameric channel, (PA₆₃)₇, formed from protective antigen (PA), which allows the other two proteins, lethal and edema factors (LF and EF), to translocate across a host cell's endosomal membrane, disrupting cellular homeostasis. It has been shown that (PA₆₃)₇ incorporated into planar phospholipid bilayer membranes forms a channel capable of transporting LF and EF. Protein translocation through the channel is driven by a proton electrochemical potential gradient on a time scale of seconds. A paradoxical aspect of this is that although LF_N (the N-terminal 263 residues of LF), on which most of our experiments were performed, has a net negative charge, it is driven through the channel by a cis-positive voltage. We have explained this by claiming that the (PA₆₃)₇ channel strongly disfavors the entry of negatively charged residues on proteins to be translocated, and hence the aspartates and glutamates on LF_N enter protonated (i.e., neutralized). Therefore, the translocated species is positively charged. Upon exiting the channel, the protons that were picked up from the cis solution are released into the trans solution, thereby making this a proton–protein symporter. Here, we provide further evidence of such a mechanism by showing that if only one SO₃[−], which is essentially not titratable, is introduced at most positions in LF_N, through the reaction of an introduced cysteine residue at those positions with 2-sulfonato-ethyl-methanethiosulfonate, voltage-driven LF_N translocation is drastically inhibited. We also find that a site that disfavors the entry of negatively charged residues into the (PA₆₃)₇ channel resides at or near its Φ-clamp, the ring of seven phenylalanines near the channel's entrance.

INTRODUCTION

The toxin produced by *Bacillus anthracis*, the causative agent of anthrax, has proved to be a promising model system to study the molecular driving forces that govern protein translocation across biological membranes. It is composed of three separate monomeric proteins: protective antigen (PA; 83 kD), named for its effectiveness in inducing protective antibody-mediated immunity against anthrax, and two enzymes, lethal factor (LF; 90 kD) and edema factor (EF; 89 kD). The three proteins may be produced recombinantly in soluble form and studied independently (for a review of anthrax toxin see Young and Collier, 2007). Toxin action is initiated by a self-assembly process. First, PA binds to a cell surface receptor, where it is activated by a furin family protease that cleaves off a 20-kD N-terminal fragment. The remaining 63-kD receptor-bound portion (PA₆₃) self-assembles into a ring-shaped homoheptamer, called

the prepore. The prepore may then form complexes with up to three molecules of EF and/or LF. These complexes are endocytosed and delivered to an acidic vesicle compartment, where the prepore undergoes an acidic pH-dependent conformational rearrangement to form a cation-selective, ion-conducting channel. The (PA₆₃)₇ channel spans the membrane as an extended, 14-stranded β barrel (Benson et al., 1998; Nassi et al., 2002; Nguyen, 2004; Katayama et al., 2008) that serves as the conduit for LF and EF into the cytosol. Once in the cytosol, LF, a Zn²⁺ protease, specifically cleaves and inactivates MAP kinase kinases (Pannifer et al., 2001), and EF, a Ca²⁺ calmodulin-dependent adenylyl cyclase, increases cAMP levels (Drum et al., 2002); both activities benefit invading *B. anthracis* by suppressing the host's immune functions (Young and Collier, 2007).

A model of the (PA₆₃)₇ 14-strand β barrel reveals that its lumen is ~15-Å wide and is able to accommodate secondary structure only as large as an α-helix (Krantz et al., 2004). Translocation through the lumen thus requires the substrates to unfold. Thermodynamic analysis

D. Basilio and S.J. Juris contributed equally to this work.

Correspondence to Daniel Basilio: dbseyler@aecom.yu.edu

S.J. Juris's present address is Depts. of Biology and Chemistry, Central Michigan University, Mount Pleasant, MI 48859.

Abbreviations used in this paper: EF, edema factor; LF, lethal factor; MTS-ACE, 2-aminocarbonyl-ethyl-methanethiosulfonate; MTS-ES, 2-sulfonato-ethyl-methanethiosulfonate; MTS-ET, 2-trimethylammonium-ethyl-methanethiosulfonate; PA, protective antigen; PEG, polyethylene-glycol; WT, wild-type.

© 2009 Basilio et al. This article is distributed under the terms of an Attribution–Noncommercial–Share Alike–No Mirror Sites license for the first six months after the publication date (see <http://www.jgp.org/misc/terms.shtml>). After six months it is available under a Creative Commons License (Attribution–Noncommercial–Share Alike 3.0 Unported license, as described at <http://creativecommons.org/licenses/by-nc-sa/3.0/>).

of the conformational stabilities of the homologous ~ 250 -residue amino-terminal domains of LF and EF (LF_N and EF_N , respectively) indicates that acidic conditions encountered in the endosome are sufficient to destabilize the native structures of these proteins (Krantz et al., 2004). When added to planar lipid bilayer membranes, $(PA_{63})_7$ forms channels with a unitary conductance of ~ 55 pS in 100 mM KCl at pH 5.5 (Krantz et al., 2005); at pM concentrations, several thousand channels can be incorporated into the membrane (Blaustein et al., 1989). It has been demonstrated that these channels are capable of translocating LF and EF across planar lipid bilayers (Krantz et al., 2006), where the amino terminus precedes the carboxy terminus into the pore (Zhang et al., 2004a), and the driving force comes from the gradient of the electrochemical potential of H^+ (Zhang et al., 2004b; Krantz et al., 2006). However, translocation is more complicated than acid-induced unfolding, followed by electrophoresis through a passive pore (Finkelstein, 2009). Both the second law of thermodynamics and the laws of electrostatics require that the translocated species have a net positive charge, given that a cis-positive voltage drives translocation, yet LF_N has a net negative charge. (LF_N contains 54 acidic side chains [19 aspartates and 35 glutamates] and 47 basic side chains [5 arginines, 9 histidines, and 33 ly-

sines] [Bragg and Robertson, 1989]. If the pKs of the side chains are the same as those of their respective free amino acids, LF_N bears a net negative charge of ~ -7 at pH 5.5 and ~ -12 at pH 6.5, a symmetric pH at which LF_N translocation can still be driven by voltage [Krantz et al., 2006]. Thus, unless the average pKs of the aspartates and glutamates on LF_N are shifted 1.5 units or more above their free amino acid values, LF_N bears a net negative charge at at least some of the pHs at which its translocation has been studied.) The simplest mechanism to achieve a positive charge on the translocated species is for the carboxyl groups of the aspartic and glutamic acid residues to be largely neutralized, i.e., protonated. Thus, these carboxyls on LF_N must pick up protons from the cis solution at some point during their entrance into the channel, and then discharge them into the trans solution as they exit the channel. In other words, the $(PA_{63})_7$ channel functions as a proton-protein symporter (Krantz et al., 2006; Finkelstein, 2009). At all times, the portion of LF_N that lies within the channel bears a net positive charge.

At pH 5.5, a carboxyl on LF_N spends $\sim 97\%$ of its time in its ionized, negatively charged form (assuming the pKs of the aspartates and glutamates on LF_N are approximately the same as those of the free amino acids). Why, then, should its minority, protonated neutral form be

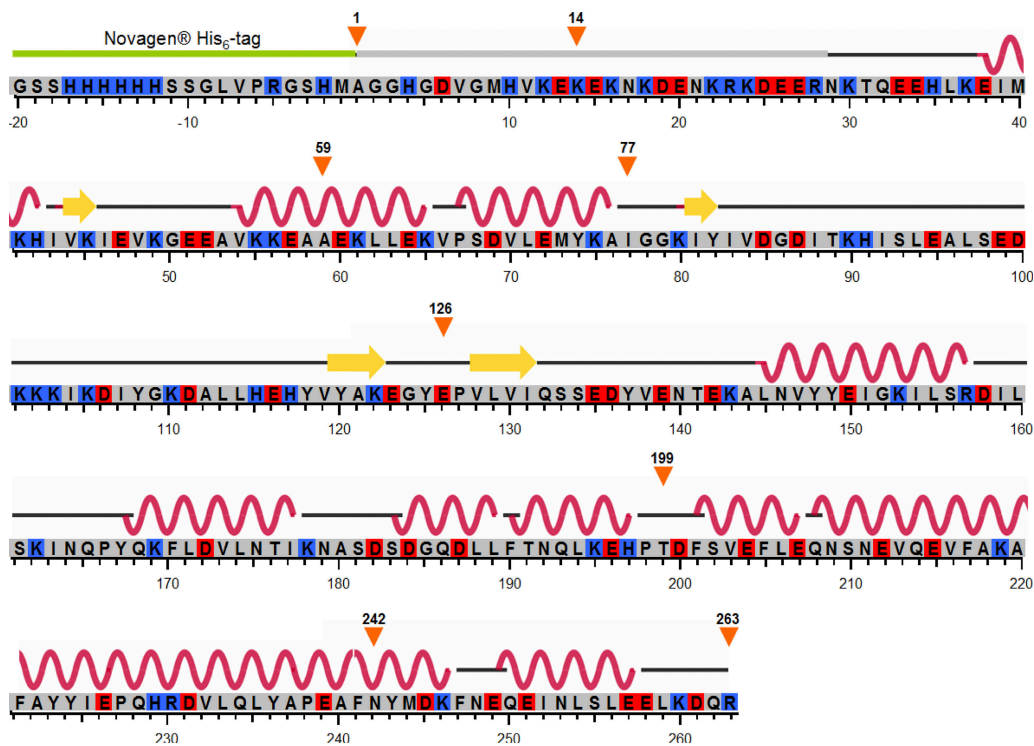


Figure 1. The primary sequence of LF_N (Bragg and Robertson, 1989) with its N-terminal Novagen His₆-tag. The orange arrows mark the cysteine mutants investigated in this paper. Acidic residues (aspartates and glutamates) are colored in red, and basic residues (lysines, arginines, and histidines) are in blue. Above the sequence, the red sine waves designate α -helices, the yellow arrows are β -structure, the black lines mark neither α -helices nor β -structure, as determined from the crystal structure of LF (PDB entry 1JKY), and the gray line designates the unresolved unstructured region, residues 1–28.

the one that enters the channel? It can be argued that the channel's significant (although not ideal [Blaustein et al., 1989; Blaustein and Finkelstein, 1990]) cation selectivity implies that it strongly disfavors entry of negatively charged side chains into the channel. Here, we provide further evidence for the importance of the protonation of the carboxyls on LF_N by showing that if only one SO₃⁻, which is essentially not titratable, is introduced at most positions in LF_N (through the reaction of an introduced cysteine residue at those positions with 2-sulfonato-ethyl-methanethiosulfonate [MTS-ES]), LF_N translocation is drastically inhibited.

MATERIALS AND METHODS

Protein Purification and Labeling with MTS Reagents

Wild-type (WT) PA (83 kD), its F427A mutant, LF_N (residues 1–263 of LF, containing the N-terminal Novagen His₆-tag), and its cysteine mutants (Fig. 1) were expressed recombinantly and purified as described previously (Krantz et al., 2004, 2005; Pimental et al., 2004; Zhang et al., 2004b). The heptameric prepore form of PA₆₃ was prepared by nicking PA₈₃ with trypsin and purifying the PA₆₃ heptamer from the smaller 20-kD fragment using anion-exchange chromatography (Cunningham et al., 2002). The MTS-labeling reaction of LF_N cysteine mutants was performed by incubation of 100 μl LF_N (2 mg/ml) with 3 μl of MTS reagents (20 μg/μl; Toronto Research Chemicals) for 30 min at room temperature in degassed 150 mM NaCl and 20 mM NaPO₄, pH 7.2. The reaction was stopped by dialyzing out the MTS reagents at 4°C. To confirm that virtually all of the protein was labeled, MTS-reacted and unreacted protein were incubated with 600 μM maleimide 5,000 polyethylene-glycol (PEG; Sigma-Aldrich) for 30 min at 30°C and then run on an SDS non-reducing acrylamide gel. The unreacted protein suffers a shift on the gel, whereas the MTS-reacted LF_N is unaffected (Fig. 2); WT LF_N, which has no native cysteines, is also unaffected (not depicted).

Planar Lipid Bilayers

Bilayers were formed by the brush technique (Mueller et al., 1963) across a 350-μm diameter aperture in a Teflon partition. Membranes separated two compartments of 3.5 ml containing symmetric buffered solutions of 100 mM KCl, 5 mM potassium succinate, and 1 mM EDTA, pH 5.5, which could be stirred by small magnetic bars. Agar salt bridges (3 M KCl, 3% agar) linked Ag/AgCl electrodes in saturated KCl baths to the cis and trans compart-

ments. The membrane-forming solution was 3% diphtanoyl-phosphatidylcholine (Avanti Polar Lipids, Inc.) in *n*-decane, and membrane formation was monitored both visually and by capacitance (~500 pF). All experiments were performed under voltage clamp conditions with a Bilayer Clamp BC-525C (Warner Instruments). Voltages are those of the cis solution (to which protein was added) with respect to the trans solution, which was held at virtual ground. Current responses were filtered at 1 kHz by a low-pass eight-pole Bessel filter (Warner Instruments), recorded by computer via an analogue-to-digital converter (NI USB-6211; National Instruments) at 20 Hz, using IGOR NIDAQ Tools MX 1.0 and IGOR 6.0.3.1 (WaveMetrics Inc.), and confirmed by a chart recorder (DMP-4B Physiograph; Narco Bio-Systems Inc.).

(PA₆₃)₇ Channel Formation, LF_N Conductance Block, and Translocation

After membrane formation, PA₆₃ prepore heptamer was added to the cis compartment (to a final concentration of ~1 ng/ml [~2 pM]), which was held at a Δψ of +20 mV with respect to the trans compartment. After the conductance consequent to (PA₆₃)₇ channel formation had reached a more-or-less constant level, LF_N was added to the cis compartment (final concentration, ~3 nM). The progress of LF_N binding to (PA₆₃)₇ channels and blocking them was monitored by the continuous fall of conductance. In most experiments, >95% of the conductance was blocked by LF_N at +20 mV before translocation experiments were begun. After LF_N conductance block of (PA₆₃)₇ channels was complete, excess ligand was removed from the cis compartment by perfusion, using a BPS-2 Bilayer Perfusion System (Warner Instruments) coupled to a peristaltic pump (Buchler) at a rate of 10 ml/min. The exchange of 10 volumes was accomplished in three and a half minutes, while Δψ was held constant at +20 mV. After this, Δψ was stepped to +55 mV, and the rate of LF_N translocation was determined, as reflected in the rate of conductance rise caused by LF_N traversing the channel and exiting into the trans solution.

In Figs. 3–6, what is plotted is the normalized conductance versus time. With the exception of the MTS-ES-reacted LF_N cysteine mutants that never reached a constant value, conductances were normalized to the level obtained after unblocking at +55 mV was completed. In general, this conductance level was ≥90% of the conductance level before blocking by LF_N. For the MTS-ES-reacted LF_N mutants, after the 55 mV was applied for the length of time indicated in the figures, the voltage was stepped for 30 s to -80 mV, and then back to +20 mV. Negative voltages drive LF_N out of the channel back to the cis side, so that the conductance immediately obtained at +20 mV (before the slow reblocking of the channels by still-attached LF_N) represents that of unblocked channels. Conductances were normalized to this level, which was ≥90% of the conductance level before blocking by LF_N.

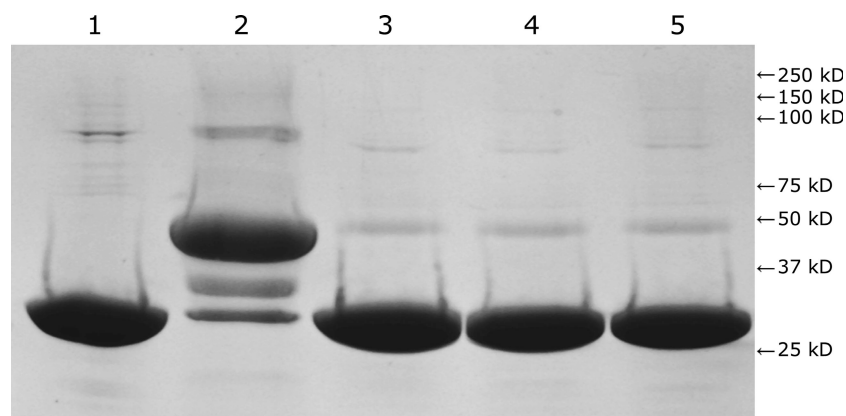


Figure 2. Example of an SDS non-reducing acrylamide gel of an LF_N cysteine mutant (I77C) reacted with MTS reagents, followed by incubation with maleimide 5,000 PEG. Lane 1, I77C; lane 2, I77C plus maleimide PEG; lane 3, I77C reacted with MTS-ES, followed by maleimide PEG; lane 4, I77C reacted with MTS-ET, followed by maleimide PEG; lane 5, I77C reacted with MTS-ACE, followed by maleimide PEG. Note the gel shift in lane 2 caused by the reaction of the maleimide PEG with the cysteine on I77C; the virtual absence of any such shift in lanes 3–5 indicates that almost all the I77C had reacted with the MTS reagent.

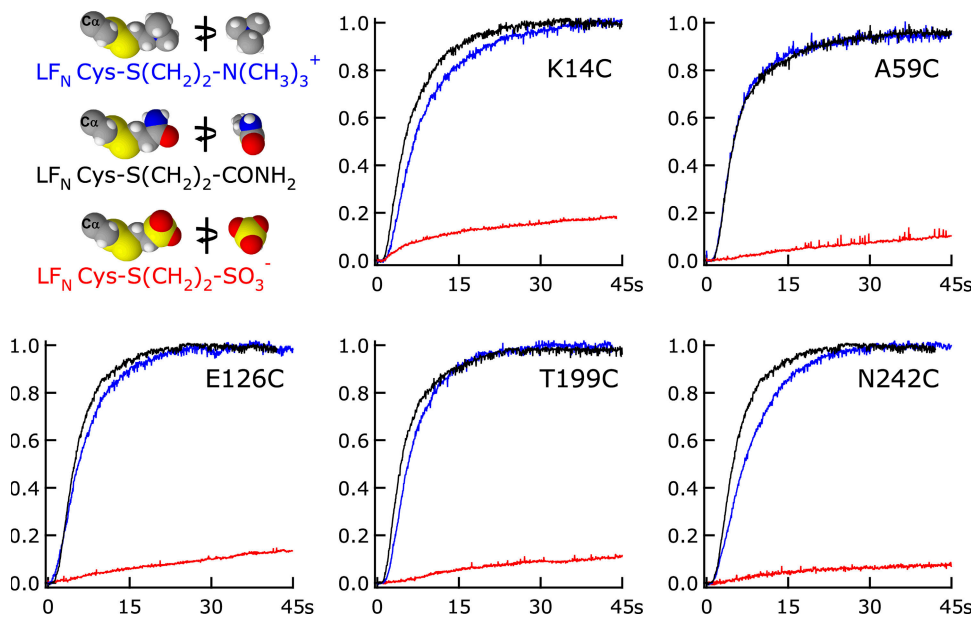


Figure 3. The effect of introducing one SO_3^- (by reacting an LF_N cysteine mutant with MTS-ES) on the translocation rate of LF_N . The panels show the normalized rise of conductance (a reflection of translocation of LF_N through the $(\text{PA}_{63})_7$ channel) that occurred when, after the perfusion of the indicated MTS-reacted LF_N cysteine mutant out of the cis compartment, the voltage was stepped from +20 to +55 mV. Note the very slow translocation of the MTS-ES-reacted LF_N (red). In contrast, the translocation rates of the MTS-ET-reacted (blue) and the MTS-ACE-reacted LF_N (black) were essentially the same as that of WT LF_N (which is not shown in the panels for clarity), with a half-time of ~ 5 s.

RESULTS

LF_N translocation is driven by a gradient of the electrochemical potential of H^+ , which can be established by either applying a $\Delta\Psi$ or a ΔpH (or both) across the membrane (see Finkelstein, 2009). Although pH-driven translocation is physiologically more relevant than voltage-driven translocation (Krantz et al., 2006), our concern here is strictly with the biophysics of voltage-driven translocation.

Effect of an Introduced SO_3^- on LF_N Translocation

Despite LF_N having a net negative charge of approximately -7 at pH 5.5, it is driven through the $(\text{PA}_{63})_7$

channel by cis-positive voltages (Zhang et al., 2004b). We have argued that this can occur because the carboxyls on the aspartates and glutamates of LF_N are, for the most part, in their protonated form within the channel (Krantz et al., 2006), and that this is because the channel strongly disfavors the entry into it of negative charges (anions) (Finkelstein, 2009). If this is so, one might expect that the introduction into LF_N of essentially non-titratable negative charges would have a detrimental effect on its translocation. And indeed, introducing only one SO_3^- at five arbitrarily chosen positions within its sequence had a profound effect on its rate of translocation. Namely, normally under

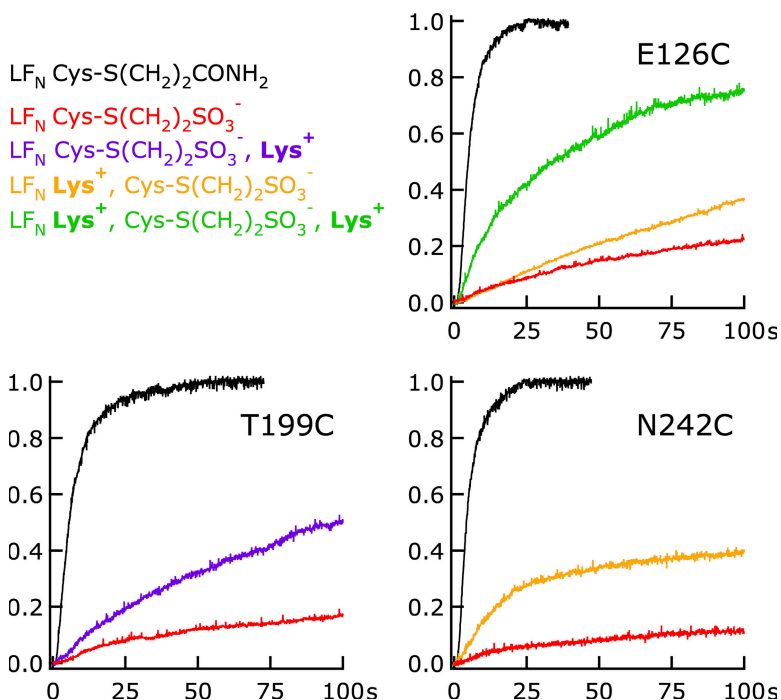


Figure 4. Partial recovery of the translocation rate of MTS-ES-reacted LF_N (red) when positively charged lysines are flanking the introduced SO_3^- . The experiments were performed as described in Fig. 3. The rate of translocation of MTS-ACE-reacted LF_N E126C, T199C, and N242C (black) was the same as that of WT LF_N and the unreacted lysine mutants (not depicted). Note that one lysine adjacent to the SO_3^- (orange in E126C and N242C, and purple in T199C) increased the rate of translocation, and two lysines, one on either side of the SO_3^- (green in E126C), produced an even greater increase in the rate of translocation, although still considerably short of the rate of translocation of WT LF_N .

our experimental conditions (symmetric pH 5.5, $\Delta\Psi = +55$ mV) the translocation half-time of LF_N (with the N-terminal Novagen His₆-tag) was ~ 5 s. However, after attaching a S-(CH₂)₂-SO₃⁻ at residue K14C, A59C, E126C, T199C, or N242C, we obtained $\leq 20\%$ translocation even after 45 or 100 s (Figs. 3 and 4). (As expected, if LF_N that had reacted with MTS-ES was subsequently treated with DTT to remove the SO₃⁻, its translocation rate was restored to normal [not depicted].) This slow translocation time is most certainly a consequence of the introduced negative charge on SO₃⁻, because attaching the similarly sized groups S-(CH₂)₂-CONH₂ or S-(CH₂)₂-N(CH₃)₃⁺ (by reacting with 2-aminocarbonyl-ethyl-methanethiosulfonate [MTS-ACE] or 2-trimethylammonium-ethyl-methanethiosulfonate [MTS-ET], respectively) had virtually no effect on the translocation rate (Fig. 3).

In support of this “electrostatic” mechanism of translocation inhibition, we found that introducing one positive charge (Lys⁺) adjacent to the SO₃⁻-bearing cysteine residue at position 126, 199, or 242 increased the rate of translocation, and introducing two positive charges (Lys⁺) flanking the SO₃⁻ at 126 (one at residue 125 and one at residue 127) further increased the rate of translocation (Fig. 4). Interestingly, the electrostatic barrier to SO₃⁻ entry into the channel is so large that even with positively charged lysines flanking the SO₃⁻ at residue 126, the half-time was considerably slower than that of WT (36 instead of 5 s; Fig. 4).

Exceptions

Whereas introducing an SO₃⁻ at most positions in LF_N drastically slowed translocation, we found two excep-

tions to this: introducing SO₃⁻ at either the C terminus (residue 263) or the N terminus (residue 1) had no effect on the translocation rate (Fig. 5). One can rationalize the former by saying that unlike an interior residue that is followed by another residue, the C-terminal residue is followed by water, which via its higher dielectric constant reduces the electrostatic barrier to entry of the SO₃⁻ into the channel. One cannot, however, make the same argument for the N-terminal residue 1 and say that it is preceded by water, because our experiments were done with the 20-residue His₆-tag attached at the N terminus. Thus, residue 1 is as much an interior residue as 14, 59, 126, 199, and 242, which, when reacted with MTS-ES, had a profound effect on translocation (Fig. 3).

Sequence gazing reveals that the neighborhood of residue 1 differs from that of the other residues we had examined in two respects: (1) it is followed by two glycines, and (2) there are positively charged residues (histidines) nearby on either side of it (at positions -2 and +4) (Fig. 1). Either one of these features could account for the failure of the introduced SO₃⁻ at position 1 to significantly slow translocation. Thus, because of their small size, the glycines could leave space for water to follow the residue; alternatively, the nearby charged residues could provide electrostatic shielding of the SO₃⁻, just as was done by the introduced lysines on either side of residue 126 (Fig. 4). A similar situation exists in the neighborhood of residue 77; namely, it too is followed by two glycines and has nearby positively charged residues (in this case lysines) on either side of it (at positions 75 and 80) at exactly the same distances as exist for residue 1. We therefore examined this position and found that if residue 77 was mutated to a cysteine and

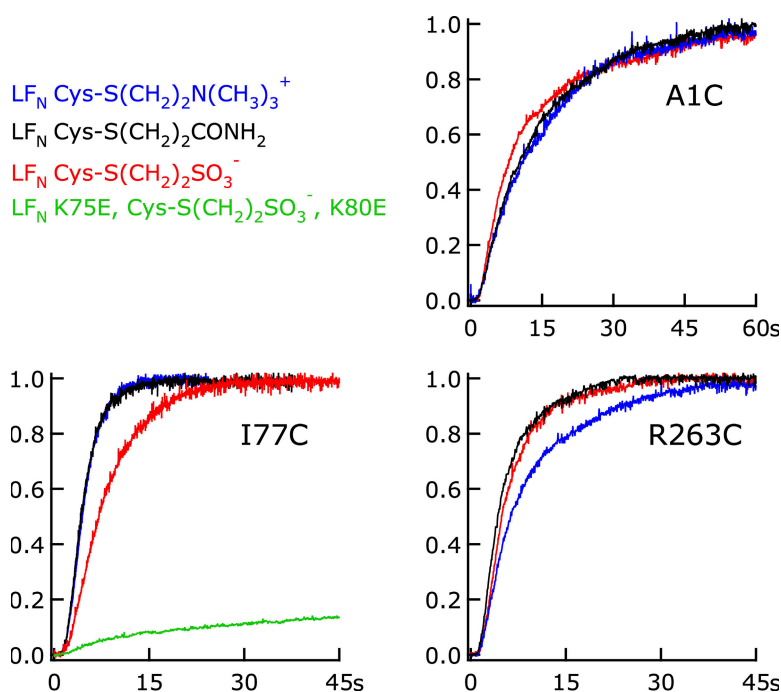


Figure 5. Introducing an SO₃⁻ at residues 1, 77, or 263 (by reacting the corresponding LF_N cysteine mutant with MTS-ES; red) had very little or no effect on the translocation rate of LF_N. The experiments were performed as described in Fig. 3. Note that if the nearby lysines were mutated to glutamates at positions K75E and K80E (green), an SO₃⁻ at position 77 greatly retarded LF_N translocation as in Fig. 4. The glutamate mutations alone did not affect the translocation rate (not depicted). (For unknown reasons, the rate of translocation of LF_NA1C reacted with MTS-ES, MTS-ET, or MTS-ACE [half-time of 10 s] was half that of WT LF_N [half-time of 5 s].)

then reacted with MTS-ES, the translocation rate was barely affected (Fig. 5). It appears that of the two possible factors that we have speculated could be the cause of this, the two glycines or electrostatic shielding, it is the latter that accounts for this at position 77, because when the lysines at positions 75 and 80 were mutated to glutamates, the introduction of an SO_3^- at position 77 now severely slowed translocation (Fig. 5). Unfortunately, mutating the histidines at positions -2 and $+4$ to aspartates had no effect on the translocation of the MTS-ES-reacted cysteine at position 1 (not depicted). Thus, why an SO_3^- at position 1 does not slow translocation remains an unexplained exception.

A Site of Negative Charge Exclusion from the Channel

Where within the channel is the negative charge exclusion of SO_3^- occurring? The $(\text{PA}_{63})_7$ channel is a mushroom-like structure with a long stem and a cap that contains the binding site for LF_N (Fig. 6 B) (Krantz et al., 2005; Katayama et al., 2008). Near the junction of the cap with the stem lies residue F427, which forms a ring of seven phenylalanines that plays an important role in protein translocation, and which we have dubbed

the Φ -clamp (Krantz et al., 2005). If the phenylalanines were mutated to alanines, protein translocation was compromised (Krantz et al., 2005, 2006). When F427 was mutated to an alanine, the rate of voltage-driven translocation decreased by about a factor of three, but, strikingly, the rate of translocation of LF_N with an attached SO_3^- enormously increased. (A much larger decrease occurs in the rate of translocation driven by a pH gradient [Krantz et al., 2005, 2006] and is probably a consequence of the creation of a leakage pathway for protons, thereby compromising the ΔpH across the Φ -clamp.) This effect of the alanine mutation is clearly seen in Fig. 6 A for $\text{LF}_N\text{A59C}$; a similar effect was seen for the other residue tested, $\text{LF}_N\text{N242C}$ (not depicted). With either WT LF_N or the neutral $(\text{CH}_2)_2\text{-CONH}_2$ attached to the $\text{LF}_N\text{A59C}$ cysteine, the half-time of translocation (at $+55$ mV) increased from ~ 5 to ~ 15 s, whereas with SO_3^- attached to the cysteine, the half-time decreased from a value $\gg 200$ to ~ 30 s. Thus, the mutation of F427 to an alanine has almost completely removed a barrier to SO_3^- translocation. The effect of the Φ -clamp on anion entry into the channel was also manifested in the channel becoming less selective for

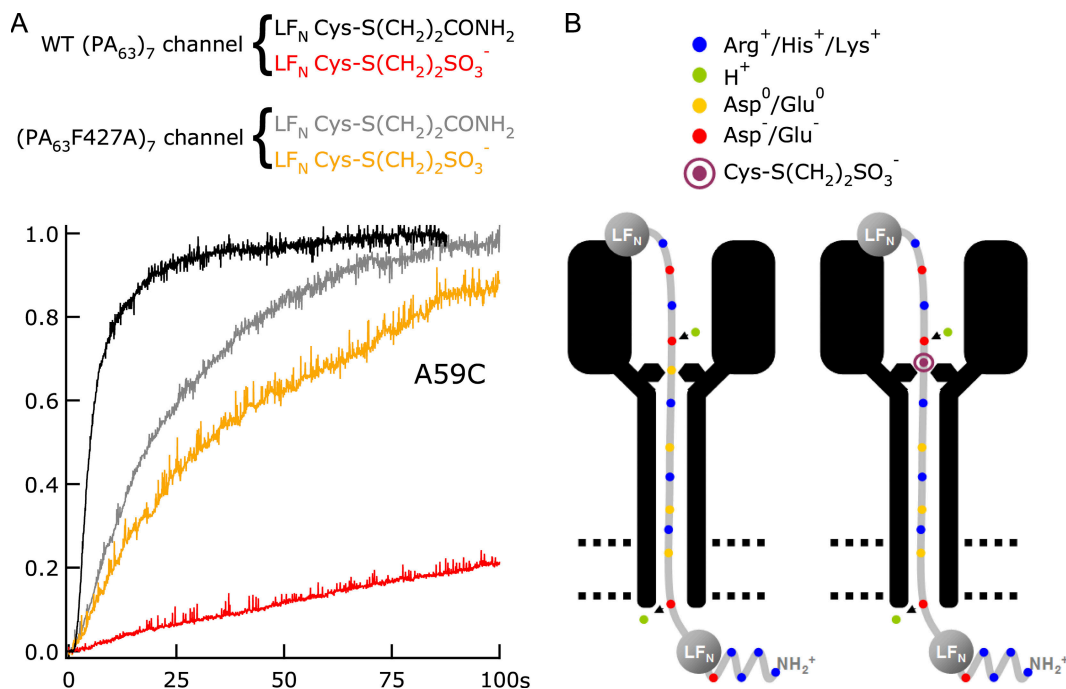


Figure 6. The effect of replacing phenylalanine with alanine at residue 427 of PA_{63} on the rate of translocation of LF_N . (A) The experiments were performed as described in Fig. 3. For the WT channel, as shown in Fig. 3 and reproduced here, the introduction of an SO_3^- at residue 59 (red) drastically reduced the rate of LF_N translocation from that of MTS-ACE-reacted LF_N (black), which was the same as that of WT LF_N (not depicted). Mutating the phenylalanine at residue 427 to alanine in the $(\text{PA}_{63})_7$ channel dramatically increased the rate of translocation of LF_N with an attached SO_3^- (orange) and decreased the rate of translocation of MTS-ACE-reacted LF_N (gray), which was the same as that of WT LF_N (not depicted). (B) Cartoon of the $(\text{PA}_{63})_7$ channel with LF_N being translocated through it. The constriction near the junction of the vestibule with the stem is the Φ -clamp, which is formed by seven F427s. We envision that for negatively charged residues (aspartates or glutamates) on LF_N to get past the Φ -clamp, they must be neutralized; that is, they pick up protons from the cis solution. This proton is subsequently discharged into the trans solution after the residue has traversed the channel. If a non-titratable SO_3^- is introduced into LF_N , it is retarded at the Φ -clamp. Mutation of F427 to alanine alters the constriction and now more readily allows a negatively charged group such as SO_3^- to get past it.

K^+ over Cl^- . We see in Fig. 7 that the $(PA_{63}F427A)_7$ channel, although still cation selective, is less so than is the WT channel.

DISCUSSION

Because LF_N translocation from the cis to the trans solution through $(PA_{63})_7$ channels is driven by cis-positive voltages, even though LF_N bears a net negative charge, we have argued that LF_N 's aspartates and glutamates must be transported in their protonated (neutralized) form (see Krantz et al., 2006; Finkelstein, 2009). That is, negatively charged residues are largely excluded at some point from entering the channel. The experiments reported here support this argument in that (with specific exceptions discussed below) the intro-

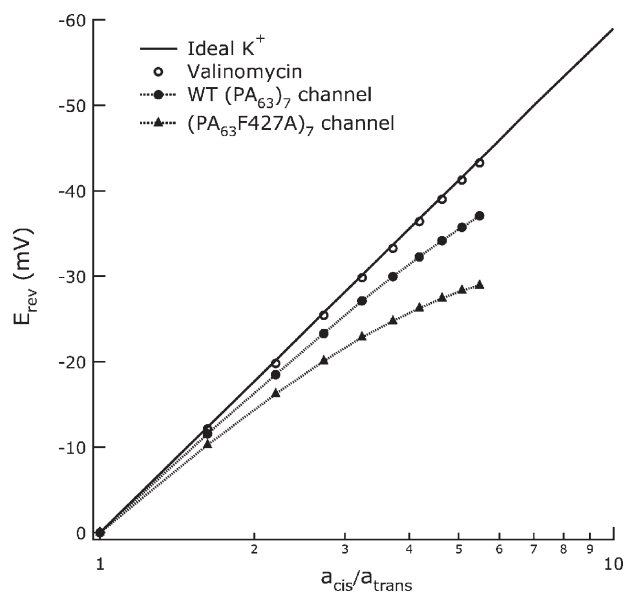


Figure 7. The effect of mutating F427 to an alanine on the ion selectivity of the $(PA_{63})_7$ channel. The experiments were started with the membrane (diphytanoyl-phosphatidylcholine) separating symmetric solutions of 100 mM KCl, 5 mM potassium succinate, and 1 mM EDTA, pH 5.5. After treating the membrane with either valinomycin (0.2 μ g/ml), WT $(PA_{63})_7$, or $(PA_{63}F427A)_7$, KCl was added in steps to the cis solution, and for each step the reversal potential (E_{rev}) was measured. Plotted in the figure is E_{rev} versus the activity ratio of KCl (a_{cis}/a_{trans}). (Activity coefficients were obtained from Appendix 8.10, Table II in Robinson and Stokes [1965], where we took the KCl concentration to be equal to the K^+ concentration. Approximately 10 mM of K^+ was contributed by K-succinate and K-EDTA.) We see in the figure that, as expected, the valinomycin-treated membrane (open circles) displayed ideal K^+ selectivity (continuous line). The selectivity of the WT $(PA_{63})_7$ -treated membrane (filled circles) deviated somewhat from ideal cation selectivity, and that of the $(PA_{63}F427A)_7$ -treated membrane (filled triangles) deviated even more so; that is, it was more permeable to Cl^- than was the WT channel. The dotted lines are drawn to connect the points. (Applying inappropriately, as usual, the Goldman-Hodgkin-Katz equation to the data, we calculate that P_K/P_{Cl} is 21 and 7.6 for the WT $(PA_{63})_7$ and $(PA_{63}F427A)_7$ channels, respectively.)

duction (by reacting single-site cysteine mutants with MTS-ES) of only one, essentially non-titratable negative charge (SO_3^-) at arbitrarily chosen positions in LF_N drastically reduced the rate of its translocation (Fig. 3). The electrostatic basis for this exclusion of SO_3^- is reinforced by our observation that it could be partially ameliorated by the placement of positive charges (lysines) adjacent to it in the amino acid sequence (Fig. 4).

The exceptions to the inhibitory effect on LF_N translocation by an SO_3^- attached to the C terminus (residue 263) or to residue 77 can be interpreted in terms of the electrostatic forces acting to exclude SO_3^- from the channel. Thus, the failure of an SO_3^- at the C terminus to reduce the translocation rate can be attributed to the water molecules surrounding SO_3^- at this position, which, because of their high dielectric constant, attenuate the electrostatic repulsive forces acting to exclude SO_3^- from the channel. An interior SO_3^- -attached residue will in general not have sufficient space around it to admit many water molecules. The failure of an SO_3^- at residue 77 to reduce translocation can be attributed to electrostatic shielding of its charge by nearby positively charged lysines at positions 75 and 80. Indeed, when they were mutated to glutamates, the introduction of an SO_3^- at position 77 drastically slowed translocation (Fig. 5). This result, obtained by removing positive charges nearby the SO_3^- and thereby slowing LF_N translocation, nicely complements the results obtained by adding positive charges nearby the SO_3^- , and thereby accelerating LF_N translocation (Fig. 4). The one exception that remains unexplained is why the introduction of an SO_3^- at position 1 (which is actually an interior residue because it is preceded by a 20-residue His₆-tag) did not retard translocation.

We also identified the Φ -clamp as a significant factor in the exclusion of SO_3^- from the channel. Thus, the $(PA_{63}F427A)_7$ channel translocated MTS-ES-reacted LF_NA59C at least an order of magnitude faster than did the WT $(PA_{63})_7$ channel (Fig. 6 A), and its selectivity for K^+ over Cl^- was diminished from that of the WT channel (Fig. 7). (The Φ -clamp is not the only site that acts to exclude Cl^- from the channel. One expects that the rings of negative charge within the channel created by residues D276, E302, E308, D315, D335, and E343 should participate in this process. And indeed, the triple-mutant channel E302S/E308S/D315S was much less selective for K^+ over Cl^- than was the F427A channel [unpublished data], and, as expected, LF_N translocation was severely impaired [unpublished data].) Why the mutation of F427 to an alanine results in a channel that is less exclusionary of anions is not obvious. One possibility is that the mutation causes a repositioning further away from the entrance to the channel of the nearby negatively charged aspartates at positions 425 and 426. It is interesting to note that this loss in anion exclusion is accompanied by a decrease in the translocation rate

of WT LF_N (Fig. 6 A). This is to be expected, because insofar as ionized glutamates and aspartates on LF_N are allowed to enter the channel, they will act to retard its (positive) voltage-driven translocation. At this time, it is not clear to what extent the slower voltage-driven translocation of LF_N through the (PA₆₃F427A)₇ channel is a consequence of its allowing negatively charged glutamates and aspartates to enter the channel, or to its effect on the unfolding of LF_N (Krantz et al., 2005).

We thank Dr. Paul Kienker for the GHK calculations and insightful comments, and Dr. Myles Akabas for critically reading this paper.

This work was supported by National Institutes of Health grants GM-29210 (to A. Finkelstein) and AI-22021 (to R.J. Collier).

Lawrence G. Palmer served as editor.

Submitted: 25 November 2008

Accepted: 15 January 2009

REFERENCES

- Benson, E.L., P.D. Huynh, A. Finkelstein, and R.J. Collier. 1998. Identification of residues lining the anthrax protective antigen channel. *Biochemistry*. 37:3941–3948.
- Blaustein, R.O., and A. Finkelstein. 1990. Voltage-dependent block of anthrax toxin channels in planar phospholipid bilayer membranes by symmetric tetraalkylammonium ions. Effects on macroscopic conductance. *J. Gen. Physiol.* 96:905–919.
- Blaustein, R.O., T.M. Koehler, R.J. Collier, and A. Finkelstein. 1989. Anthrax toxin: channel-forming activity of protective antigen in planar phospholipid bilayers. *Proc. Natl. Acad. Sci. USA*. 86:2209–2213.
- Bragg, T.S., and D.L. Robertson. 1989. Nucleotide sequence and analysis of the lethal factor gene (*lef*) from *Bacillus anthracis*. *Gene*. 81:45–54.
- Cunningham, K., D.B. Lacy, J. Mogridge, and R.J. Collier. 2002. Mapping the lethal factor and edema factor binding sites on oligomeric anthrax protective antigen. *Proc. Natl. Acad. Sci. USA*. 99:7049–7053.
- Drum, C.L., S.Z. Yan, J. Bard, Y.Q. Shen, D. Lu, S. Soelaiman, Z. Grabarek, A. Bohm, and W.J. Tang. 2002. Structural basis for the activation of anthrax adenyl cyclase exotoxin by calmodulin. *Nature*. 415:396–402.
- Finkelstein, A. 2009. Proton-coupled protein transport through the anthrax toxin channel. *Philos. Trans. R. Soc. Lond. B Biol. Sci.* 364:209–215.
- Katayama, H., B.E. Janowiak, M. Brzozowski, J. Juryck, S. Falke, E.P. Gogol, R.J. Collier, and M.T. Fisher. 2008. GroEL as a molecular scaffold for structural analysis of the anthrax toxin pore. *Nat. Struct. Mol. Biol.* 15:754–760.
- Krantz, B.A., A.D. Trivedi, K. Cunningham, K.A. Christensen, and R.J. Collier. 2004. Acid-induced unfolding of the amino-terminal domains of the lethal and edema factors of anthrax toxin. *J. Mol. Biol.* 344:739–756.
- Krantz, B.A., R.A. Melnyk, S. Zhang, S.J. Juris, D.B. Lacy, Z. Wu, A. Finkelstein, and R.J. Collier. 2005. A phenylalanine clamp catalyzes protein translocation through the anthrax toxin pore. *Science*. 309:777–781.
- Krantz, B.A., A. Finkelstein, and R.J. Collier. 2006. Protein translocation through the anthrax toxin transmembrane pore is driven by a proton gradient. *J. Mol. Biol.* 355:968–979.
- Mueller, P., D.O. Rudin, H.T. Tien, and W.C. Westcott. 1963. Methods for the formation of single bimolecular lipid membranes in aqueous solution. *J. Phys. Chem.* 67:534–335.
- Nassi, S., R.J. Collier, and A. Finkelstein. 2002. PA₆₃ channel of anthrax toxin: an extended beta-barrel. *Biochemistry*. 41:1445–1450.
- Nguyen, T.L. 2004. Three-dimensional model of the pore form of anthrax protective antigen. Structure and biological implications. *J. Biomol. Struct. Dyn.* 22:253–265.
- Pannifer, A.D., T.Y. Wong, R. Schwarzenbacher, M. Renatus, C. Petosa, J. Bienkowska, D.B. Lacy, R.J. Collier, S. Park, S.H. Leppla, et al. 2001. Crystal structure of the anthrax lethal factor. *Nature*. 414:229–233.
- Pimental, R.A., K.A. Christensen, B.A. Krantz, and R.J. Collier. 2004. Anthrax toxin complexes: heptameric protective antigen can bind lethal factor and edema factor simultaneously. *Biochem. Biophys. Res. Commun.* 322:258–262.
- Robinson, R.A., and R.H. Stokes. 1965. *Electrolyte Solutions*. Butterworth, London. 494 pp.
- Young, J.A., and R.J. Collier. 2007. Anthrax toxin: receptor-binding, internalization, pore formation, and translocation. *Annu. Rev. Biochem.* 76:243–265.
- Zhang, S., A. Finkelstein, and R.J. Collier. 2004a. Evidence that translocation of anthrax toxin's lethal factor is initiated by entry of its N terminus into the protective antigen channel. *Proc. Natl. Acad. Sci. USA*. 101:16756–16761.
- Zhang, S., E. Udho, Z. Wu, R.J. Collier, and A. Finkelstein. 2004b. Protein translocation through anthrax toxin channels formed in planar lipid bilayers. *Biophys. J.* 87:3842–3849.